Low temperature hydrothermal growth of aligned zinc oxide nanowires using FTO substrate

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Abstract— Zinc oxide nanowires have been mostly observed due to special material properties and showing special performance in photonic, electronics and optics. Aligned Zinc oxide nanowires were grown on FTO substrate by using the hydrothermal process. This growth technique is bottom up, low cost and easy. The seed layer is deposited onto the substrate by spin coating with low temperature and nanowires are affected by the thickness of seed layer. Afterwards, the nanowires were growing hydrothermally onto those seed layers. Different characterizations have been done in detail using X-ray Diffraction (XRD), Scanning electron microscopy (SEM), Raman spectroscopy and photoluminescence spectroscopy (PL). The SEM is used to relives the presence of nanowire with hexagonal shape in vertical direction. The XRD measurements show diffraction pattern of wurtzite structure.

Keywords-ZnO, FTO, nanowires, hydrothermal growth

I. Introduction

The ZnO has a wide band gap semiconductor due to which it become best candidate among different wide band gap materials for large scale and economic optoelectronics applications [1, 2]. ZnO has high radiation resistance, high break down fields, large situation rate, effective luminescence and so that's why it has utilized for high power devices at high frequency and high temperature [3]. Zinc oxide has various advantages over remaining compound semiconductors in manufacturing costs which are less as compare for GaN. Zinc oxide has a large exciton binding energy range up to 60 mev which is much greater than ZnSe (26 mev) and GaN (25 mev), this make them stable for Ultra Violet emitter at the room temperature and also a best candidate for fabrication of blue and UV LEDs [4]. Zinc oxide nanowires (NWs) have one-dimensional (1-D) structure showing quantum confinement effect and also having a greater surface to volume ratios. ZnO has a n-type semiconductor material and have difficult task for p-type doping. ZnO has a wurtzite crystalline structure. Zinc oxide has synthesized to nanowire by different methods such as vapor-liquid-solide growth [5, 6], chemical vapour deposition [7-9], pulse laser ablation techniques [10], and hydrothermal method. On different temperature ZnO nanowires has been synthesized, as vapor phase process done on high temperature and

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Zahid Ali NILOP, Pakistan solution phase require low temperature. Initially the ZnO nanowires were synthesized by vapor phase owing high vacuum and high temperatures (450 to 900°C) [11]. But it limit the integration of flexible electronic devices and large size of substrate. Other popular method for synthesis of ZnO nanowires is vapor liquid solid growth, settling high quality nanowire using sapphire as a substrate [12]. ZnO nanowires also grow by pulse laser deposition method. Usually high quality of ZnO nanowires produce by the gas phase, however the commercial potential is restrict by the high temperature round about 500°C, for allied array sapphire substrate is use and also high cost of deposition. Only two important approaches for low temperature methods remain for growth is hydrothermal method and electrochemical method [13]. Low temperature growth is promising candidate for commercialization due their large scale production, their simplicity, high quality products and low cost. Some advantages of hydrothermal process over the remaining method is that its low temperature (90°C) growth, work at atmospheric pressure, cover lager area of substrate about 4inch diameter, easy to control and environmental friendly. Smooth and uniform growth of ZnO nanowire array with high aspect ratio is synthesized by low temperature hydrothermal method. Their amazing performance in photonics, optics and electronics ZnO NWs become much appealing candidates for multiple uses such as light-emitting diodes [14], gas sensors [15], UV lasers [16], solar cells [17], photo-detectors [18], resistive switching [19] and photo-catalysts [20].

п. Experimental

A. ZnO seed solution preparation

By using sol-gel spin coating method a uniform and homogenous zinc oxide nanoparticles were deposited. Before the deposition of seed layer, solution was put together by utilizing zinc acetate dehydrate [Zn (CH₃COO)₂. 2H₂O] as a precursor and Ethanol (EtOH) use as a solvent. Zinc oxide solution was obtained by mixing the solvent with zinc acetate dehydrate. To confirm the zinc power was fully mixed in the ethanol, the mixture was stirring constantly for 1 h without heating. Before deposition the solution was stored in room temperature for 24 h.

B. ZnO seed layer preparation

A FTO coated glass template was used as a substrate. Before the seed layer coating on substrate, the substrate was passed through a standard cleaning process. It was cleaned in ultrasonic bath with acetone, poly ethylene glycol and ethanol and then rinsed with deionized water (DIW). In last, the substrate was dried with optical tissue. A spin coater was used for deposition of zinc oxide solution on sterilized substrates at 3,000 rpm for 30 s. After coating, the substrate



was dried on hot plate for round about at 100°C for 5 minutes. The process of deposition was repeated four times to get more homogenous and thicker ZnO films. To remove the solvent and organic components from films, the annealing process was done. Conventional furnace was used for annealing purpose. Figure 1 shows the preparation process of zinc oxide thin film.



Figure 1: ZnO thin film preparation process flow.

c. ZnO NWs formation

Zinc oxide nanowires were obtained from the uniform coated ZnO nanoparticles on substrate by hydrothermal growth. The homogenous solution consisted of hexamethylenetetramine (HMT) and zinc nitrate hexahydrate. The complete dissolution of HMT powder and Zn $(NO_3)_2$ in DIW was done with the help of stirrer for 15 minutes at room temperature. The nanowires were grown by dipping the substrate that was settled upside down in above solution. The growth procedure was done in laboratory oven by heating the solution at 90°C for 3 h. After the growth of nanowires, sample washed with tape water and then rinsed with DIW for elimination of salts from sample surface. Last step, the sample placed in oven for heat-treated for 1/2 h at 300°C. The growth procedure of zinc oxide nanowires is shown in figure 2.



Figure 2: ZnO nanowires growth process.

ш. Material Characterization

By using scanning electron microscopy surface morphology of zinc oxide nanowires was analyzed. Crystal properties and crystallization was study by X-ray diffraction. Through photoluminescence the luminescence and optical properties of zinc oxide nanowire were studied. In addition, Raman provides a fingerprint of molecule by which it can be identified.

IV. Result and discussion

A. SEM Characterization

Figure 3(a), (b) show the top and cross-sectional view of zinc oxide nanowires by using SEM that was synthesized by using ethanol as a solvent. Hexagonal morphology was showed by all the synthesized zinc oxide nanowires. Entire substrate was covered by nanowires, and an aliened nanowire was mostly grown. The diameter of zinc oxide nanowires was obtained approximately 200nm. Mostly nanowires showed largest diameter which synthesized by ethanol as a solvent. The diameter of nanowires is strongly influenced on the grain size of zinc oxide seed layer. As the grain size increased, the diameters of nanowires were also increase.







Figure 3: SEM images of ZnO nanowires on FTO (a) 3 times coated (b) 4 times coated, taken at 20.0KV (Inset) Cross-sectional view of ZnO nanowires

B. XRD Characterization

The microstructure and crystal structure were examined by using X-Ray Diffraction. Figure 4 exhibit the patterns of zinc oxide nanowires using XDR that were synthesized on FTO substrate with seed layer and with aqueous solution. The range of peak intensity that was measured was 30° to 70° at 20. The zinc oxide nanowire that was synthesized by the hydrothermal method showed the result with intense diffraction peak at (002) plane, that is detected in between 34° to 35°. This result shows the pure hexagonal wurtzite structure in c-axis orientation of ZnO samples possessed.



Figure 4: XRD spectra of ZnO nanowires on FTO.

c. PL Characterization

In this characterization we studied the effect of ethanol on luminescence properties of zinc oxide nanowires via photoluminescence spectroscopy. Figure 5 indicates the spectra of PL that grown on substrate. Sample exhibit two clear peaks, which had ultra violet emissions of 350 to 420 nm and the visible emission at 420 to 630 nm. In the UV range the first emission band was located which caused by the recombination of free exictons through an excitonexciton collision. Second emission band emerge from the deep-level emission. This band exhibits the radiative recombination of photogenerated hole with electron that possessing to the individually ionized oxygen vacancies. Crystal quality of material can be evaluated by using UV luminescence and structural defects can be determined by using visible luminescence.



Figure 5: PL spectra of ZnO nanowires on FTO.

D. Raman Characterization

The Raman spectroscopy data of both ZnO nanowires samples on FTO are shown in figure 6. As evident in figure 4, the Raman spectra shows a strong peak at 435 cm^{-1} for



the ZnO nanowires of both samples, which correspond to the E 2 (high) mode of the Raman active mode and it is a characteristic peak for the wurtzite hexagonal phase of ZnO nanowires, as clear in XRD spectra and SEM images. The strong peak at 435 cm⁻¹ and no other peak in the Raman spectra for ZnO nanowires is clear proving that our samples have high crystalline quality.



Figure 6: Raman spectra of ZnO nanowires on FTO.

v. Conclusions

In this study, zinc oxide nanowires were synthesized by a convenient hydrothermal method and a low cost with highly crystalline structure. The diameter and length of hydrothermal synthesized zinc oxide nanowires is determine by the SEM image of samples. The diameters of zinc oxide nanowires range from 150 to 200 nm and length range from 1000 to 1500 nm. The XRD patterns showed that all zinc oxide nanowires had extraordinary crystal quality and having high c-axis orientations. The diameter, crystal quality, optical bandgap and grain size of zinc oxide nanowire were influenced by the variety of solvent used in preparation of seed layer. The method elaborate in this study is only to provide a low cost, low temperature and simple fabrication of zinc oxide nanowires with preferred properties. Hydrothermal growth method for zinc oxide nanowires could be helpful in large area of application ranging from resistive switching to light emitting diodes.

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