

Empirical Correlations among Liquid Limit, Clay Fraction, and Specific Surface Area for Kaolin and Calcium Bentonite Compounded Samples

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Abstract—Specific surface area (SSA) is one of the engineering properties of clayey soils. This property is highly related to the particle size, strength, compressibility, and permeability of soils. Particle size (i.e., clay fraction) is the most important physical property of clay minerals. Liquid limit (LL), as one of Atterberg's limit, is also an important index to classify fine-grained soils. Kaolinite (non-expansive soils) and calcium bentonite (expansive soils) are selected as primer soils, and 0% to 40% sand is added to each soil to reduce clay fraction. These ten representative samples are then tested using the BET method to determine the SSA. The SSA of ten compounded samples of kaolinite and calcium bentonite are in the range of 16.2 m²/g–18.8 m²/g and 58 m²/g–69 m²/g, respectively. Results also show that the increase in clay fraction is accompanied by an increase in the SSA of both kaolin and bentonite. The increase in LL is also accompanied by an increase in SSA. The empirical correlations among LL, clay fraction, and SSA are close to Dolinar and Skrabl's equation.

Keywords—BET, clay, calcium bentonite, kaolin, specific surface area, permeability

I. Introduction

Specific surface area (SSA) largely determines the engineering properties of clayey soils. This property is highly related to the shear strength, compressibility, and permeability of soils. Liquid limit (LL), as one of Atterberg's limit, and particle size (i.e., clay fraction) are the most important physical properties to classify fine-grained soils. This study determines the empirical correlation among SSA, LL, and clay fraction.

II. Methods

Kaolinite (non-expansive soils) and calcium bentonite (expansive soils) were used as primer soils, and 0% to 40% sand was added to each soil to reduce clay fraction. Ten representative samples of kaolinite and calcium bentonite (Ca-Bentonite) samples were then tested to determine SSA. As seen in Table 1, the increase in sand content was accompanied by a reduction in both clay fraction and LL.

As Fig. 1 shows, the SSA based on BET's adsorption isotherm with the gas N₂ was determined using a laboratory instrument called surface area analyzer (Quantachrome Instruments-Nova 2000). The LL of kaolinite ranged from 39.15 to 51.90. The LL of Ca-Bentonite ranged from 56.20 to 93.50. The increase in sand in the mixture of kaolin and sand was accompanied by a reduction in LL.

The clay fraction of kaolinite and Ca-Bentonite was in the range of 29%–44.7% and 13%–27%, respectively. The SSA of each compounded sample of kaolinite and Ca-Bentonite was in the range of 16.22 m²/g–18.80 m²/g and 58 m²/g–69 m²/g, respectively.



Figure 1. Determination of SSA using BET method.

The activity (A) of kaolinite soils was in the range of 0.12–0.29. According to Ishibashi and Hazarika (2015), kaolinite samples are categorized as inactive soil. However, Ca-Bentonite had an activity in the range of 0.6–0.94. This means that the latest sample was classified as inactive to normal soil.

III. Result and Discussion

A. Relationship between LL and SSA

Fig. 2 shows the relationship between SSA and LL. The increase in LL is accompanied by an increase in SSA. For both kaolinite and Ca-Bentonite, the proposed empirical equations are as follows:

$$SSA = 0.33LL + 4.33 \quad (\text{for Kaolinite}) \quad (1)$$

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$$SSA = 0.33LL + 38.91 \text{ (for Ca-Bentonite)} \quad (2)$$

$$SSA = 1.17(\%clay) + 48.33 \text{ (for Ca-Bentonite)} \quad (4)$$

As Fig. 2 shows, the SSA of Dolinar and Skrabl (2011) is higher than the results in both Eqs. (1) and (2). This difference may be due to the different types of soil and mineral compositions (Lambe and Whitman, 1969). For example, South African soils (De Bruyn et al., 1957) have higher SSA results than British soils (Farrar and Coleman, 1967) (refer to Fig. 3).

TABLE I. SOILS PARAMETERS

Sample	Liquid Limit (LL)	Clay Fraction (% Clay)	Activity (A)	Specific Surface Area (SSA) (m ² /g)
Non-Expansive Soils	100% Kaolin	51.90	0.12	16.22
	90% Kaolin + 10% Sand	46.80	0.18	18.70
	80% Kaolin + 20% Sand	43.25	0.16	17.12
	70% Kaolin + 30% Sand	41.15	0.18	18.65
	60% Kaolin + 40% Sand	39.50	0.29	18.80
Expansive Soils	100% Ca-Bentonite	93.50	0.94	69.33
	90% Ca-Bentonite + 10% Sand	85.50	0.93	68.44
	80% Ca-Bentonite + 20% Sand	71.80	0.75	66.41
	70% Ca-Bentonite + 30% Sand	67.20	0.6	58.96
	60% Ca-Bentonite + 40% Sand	56.20	0.81	58.02

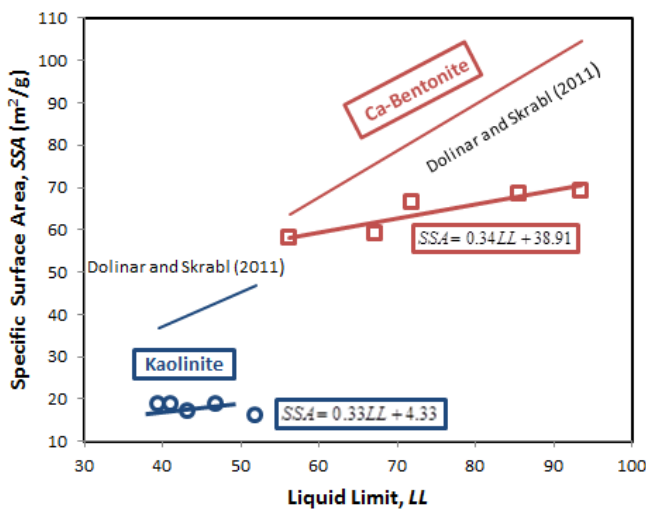


Figure 2. Comparison of Dolinar and Skrabl's method and the test results.

B. Relation between Clay Fraction and SSA

The relationship between SSA and clay fraction (%clay) also shows a similar tendency with the previous correlation. The correlations for both soils are as follows:

$$SSA = 0.33(\%clay) + 4.33 \text{ (for Kaolinite)} \quad (3)$$

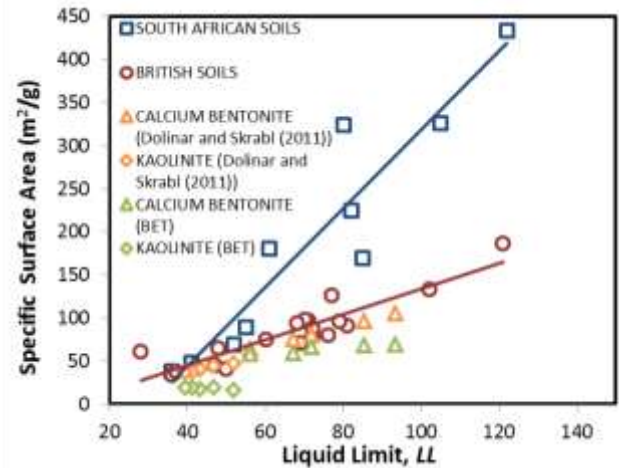


Figure 3. Comparison of SSA and LL from other scholar's results.

As Fig. 4 shows, the correlations are lower than those in Dolinar and Skrabl (2011). Table 2 shows that the reduction of clay fraction is accompanied by a reduction in LL. Hence, the trend between SSA and clay fraction has a proportional result compared with the trend between SSA and LL.

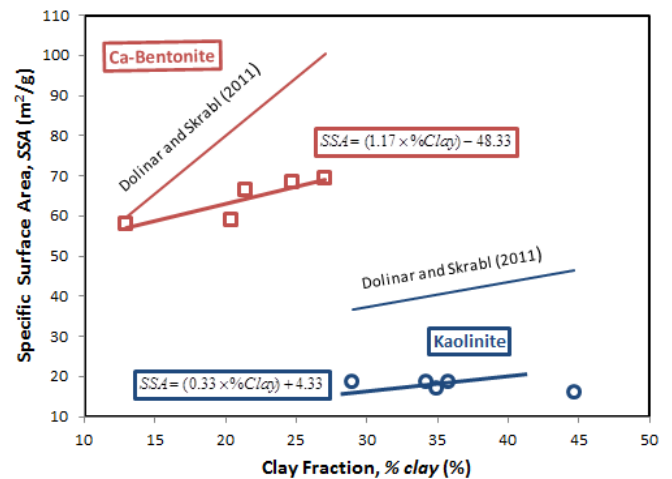


Figure 4. Relationship between clay fraction and SSA.

C. Relation Between Permeability and SSA at LL

The Kozeny and Carman method is used to predict the permeability of soil (Kozeny, 1927; Carman, 1939). The equation is as follows:

$$k = \left(\frac{\gamma}{\eta} \right) \left(\frac{1}{C_{K-C}} \right) \left(\frac{1}{S_o^2} \right) \left(\frac{e^3}{1+e} \right) \quad (5)$$

where γ is the unit weight of water, η is the viscosity of water, C_{K-C} is the Kozeny-Carman empirical equation, S_o is the SSA per unit volume of particles (1/m), and e is the void ratio.

The void ratio for this case is calculated using the specific gravity and water content at LL. The result shows that the permeability (k) of kaolinite soil is between 6.05×10^{-14} and 2.05×10^{-13} m/s. The k value of Ca-Bentonite is in the range of 1.74×10^{-14} – 3.87×10^{-14} m/s. This value is close enough to the results of Morris (2003), where k at LL is about 2.5×10^{-13} m/s. However, the k value of Ca-Bentonite is about five times smaller than the value suggested by Morris.

For SSA plotted with k , the result is shown in Fig. 5. The increase in SSA is accompanied by an increase in the k of kaolinite and Ca-Bentonite. The k value of Ca-Bentonite is smaller than that of kaolinite. This may be due to the different interparticle interactions of kaolinite and bentonite. The increase in the void ratio (e) of Ca-Bentonite may be due to the lower interparticle force. For instance, the e of Ca-Bentonite is 1.4–2.2, and that of kaolinite is 1.1–1.4.

According to Dolinar and Trauner (2004), the quantity of free water depends on the external SSA of kaolinite. However, for Ca-Bentonite, the quantity of interlayer water is independent of the internal specific surface (i.e., quantity of adsorbed water on the clay surface and quantity of interlayer water) and dependent on the type, quantity of interlayer cations, and chemical composition of pore water (Grim, 1962; Dolinar and Trauner, 2004; Widjaja, 2010).

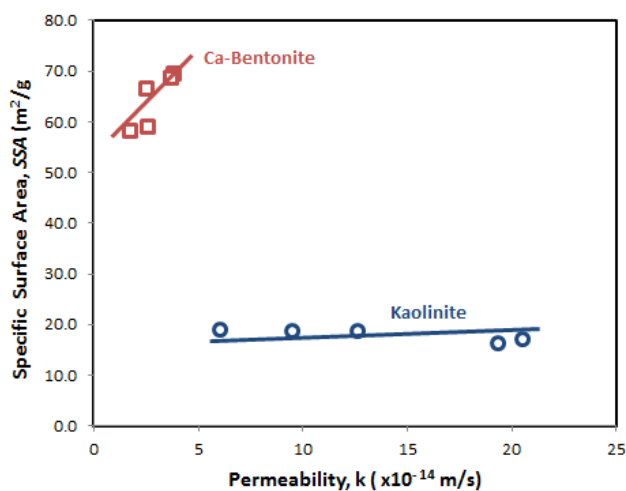


Figure 5. Different trends of the permeability of kaolinite and Ca-Bentonite soils.

IV. Conclusion

LL and clay content are linearly dependent on the SSA of both kaolinite and Ca-Bentonite. These trends are similar to Dolinar and Skrabl's result. However, the proposed correlation is lower than Dolinar and Skrabl's correlation. This difference may be due to the type and composition of soil minerals.

The permeabilities derived from the Kozeny–Carman equation for kaolinite and Ca-Bentonite at water content equal to LL are close to the permeability suggested by Morris. The increase in SSA is accompanied by an increase in permeability. The permeability of Ca-Bentonite is lower than that of kaolinite.

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